# FUROSTANOL GLYCOSIDES FROM TRIGONELLA FOENUM-GRAECUM SEEDS

## RAJESH K. GUPTA, DHARAM C. JAIN and RAGHUNATH S. THAKUR

Central Institute of Medicinal and Aromatic Plants, Lucknow 226016, India

(Revised received 11 March 1985)

Key Word Index—Trigonella foenum-graecum; Leguminosae; fenugreek; furostanol glycosides, trigofoenosides A and D.

**Abstract**—Two new furostanol glycosides trigofoenosides A and D have been isolated from the *Trigonella foenum-graecum* seeds as their methyl ethers, A-1 and D-1. Their structures have been determined as (25S)-22-O-methyl-furost-5-ene-3 $\beta$ ,26-diol, 3-O- $\alpha$ -L-rhamnopyranosyl  $(1 \rightarrow 2)$ - $\beta$ -D-glucopyranoside; 26-O- $\beta$ -D-glucopyranosyl  $(1 \rightarrow 3)$ - $\beta$ -D-glucopyranoside; 26-O- $\beta$ -D-glucopyranoside (D-1).

## INTRODUCTION

In continuation of our studies [1] on the seeds of *Trigonella foenum-graecum* saponins, we have further isolated two more new furostanol saponins and their structures have been elucidated.

#### RESULTS AND DISCUSSION

A methanolic extract of seeds was fractionated with n-butanol which yielded a crude mixture of saponins. Further separation was effected using droplet counter current and column chromatography leading to the isolation of pure methyl ethers of two saponins, designated as A-1 and D-1. The formation of methyl ethers in the case of furostanol saponins has been reported earlier [1] and for convenience these have been employed in our studies to elucidate their structures. The furostanol nature of these saponins was established through characteristic

colour reactions [2], enzymatic hydrolysis and spectral data [3].

Inspection of the Fast Atom Bombardment (FAB) mass spectrum of trigofoenoside A revealed that the molecular weight of A is 902 which was clear from the peaks at m/z1036  $[M+H+Cs]^+$ , 925  $[M+Na]^+$  and 941  $[M+K]^+$ . A prominent peak at m/z 885  $[M+H-H_2O]^+$ was observed due to loss of water involving the hydroxy function at C-22, which also suggested the furostanol structure of this saponin [4]. The peaks at m/z 723 [(M  $+H-H_2O)-162$ <sup>+</sup> and 739  $[(M+H-H_2O)-146]$ <sup>+</sup> resulted from the loss of glucose and rhamnose, respectively. The peak at  $707 [(M + H - H_2O) - 178]^+$  was assigned to the loss of one glucose with an adjacent oxygen atom indicating the outer position for one glucose unit. Two signals corresponding to m/z 577 and 561 represented the cleavage of a glucorhamnosyl (308) unit at C-3 with and without an oxygen atom. Similarly the peak at m/z 545  $[(M+H-H_2O)-340]^+$  was assigned to the

loss of one rhamnose (146+16) and one glucose (162+16) units along with their glycosidic oxygen atoms which indicated that a glucose unit was linked to the genin molecule at C-26 which was the only alternative position available for glycosidation. The peaks at m/z 415 and 397 were ascribable to the [aglycone+H]<sup>+</sup> and [(aglycone+H)-H<sub>2</sub>O]<sup>+</sup>.

Thus the sugar sequence in A could be proposed as rhamnose → glucose → yamogenin; 26-O-glucose. Acid hydrolysis of A-1 and D-1 gave the same sapogenin, yamogenin. The sugar components were D-glucose and L-rhamnose for both in the molar ratio of 2:1 for A-1 and 3:1 for D-1.

Methylation studies, periodate oxidation and partial hydrolysis suggested the sugar sequence L-rhamnopyranosyl (1  $\rightarrow$  2)-D-glucopyranose for A-1 (also supported by FAB-MS results) and L-rhamnopyranosyl (1  $\rightarrow$  2) [D-glucopyranosyl (1  $\rightarrow$  3)]-D-glucopyranose for D-1 at the C-3 position of aglycone, whereas the presence of one D-glucose at the C-26 position was confirmed by enzymatic hydrolysis for both compounds.

The anomeric configuration of D-glucose and L-rhamnose was established as  $\beta$  and  $\alpha$  respectively which was revealed by the application of Klyne's rule [6] and <sup>1</sup>H NMR spectral data. In addition enzymatic hydrolysis with a  $\beta$ -hydrolysing enzyme suggested the  $\beta$ -configuration for the C-26 glucose. Accordingly, the structure of A-1 was elucidated as (25S)-22-O-methylfurost-5-ene-3 $\beta$ ,26-diol, 3-O- $\alpha$ -L-rhamnopyranosyl(1  $\rightarrow$  2)- $\beta$ -D-glucopyranoside; 26-O- $\beta$ -D-glucopyranosyl (1  $\rightarrow$  3)]- $\beta$ -D-glucopyranoside; 26-O- $\beta$ -D-glucopyranoside for D-1.

## **EXPERIMENTAL**

Mps are uncorr. TMS was used as an internal standard in CD<sub>3</sub>OD-CDCl<sub>3</sub> and DMSO-d<sub>6</sub> for <sup>1</sup>H NMR (400 MHz and FT-80A). Column chromatography was on silica gel 60-120 mesh (BDH). Whatman No. 1 paper was used for PC. The following solvents were employed, solvent a, CHCl3-MeOH-H2O (65:40:12); solvent b, CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (65:35:10); solvent c, BuOH-pyridine- $H_2O$  (6:4:3); solvent d,  $C_6H_6$ - $Me_2CO$  (3:1); solvent e, C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO (85:15); solvent f, BuOH-EtOH-H<sub>2</sub>O (5:1:4); solvent g,  $CH_2Cl_2-Me_2CO$  (49:1); solvent h, EtOAc-C<sub>6</sub>H<sub>6</sub> (15:85). Spraying reagents, 10% H<sub>2</sub>SO<sub>4</sub>, Ehrlich's reagent and Liebermann-Burchard reagent. Sugars and methylated derivatives were located on PC (descending) by aniline hydrogen phthalate and ammoniacal AgNO<sub>3</sub> soln. GLC of sugars, dual FID; column 6', 3 % OV-17 chromosorb-W, N2 as a carrier gas, conditions for temperature programming: (a) initial hold at an initial temp. of 125° for 4 min and then at the rate of 10°/min to a final temp. of 265°; (b) same column with initial hold at an initial temp. 150° for 2 min and then at a rate of 10°/min to a final temp. of 275°. DCCC was performed using the DCC-A apparatus of Tokyo Rikakikai, Tokyo (Japan). 300 tubes were used. The solvent system used was CHCl3-MeOH-H2O (7:13:8), upper layer (water layer) as stationary phase, in descending mode. FAB-MS, JMS-D × 300 mass spectrometer.

Isolation. Fraction I of the n-BuOH extract [1] was chromatographed on a silica gel column with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (65:15-35:10). Five major furostanol glycosides, trigofoenosides A-1 to E-1 were isolated in the order of increasing polarity. Trigofoenosides A-1 and D-1 thus obtained were purified by DCCC. Samples were dissolved in 10 ml mixture (1:1) of both upper and lower phases and then chromatographed in a 10 ml

sample column. The flow rate was 7-10 ml/hr. The eluents were collected in 5 ml fractions, monitored by TLC, with solvent system b.

Trigofoenoside A-1. An amorphous solid from MeOH-Me<sub>2</sub>CO,  $R_f$  0.84 (solvent system a), mp 210-213° (decomp.); IR ν KBr cm<sup>-1</sup>; 3600-3200 (OH), 1150-1000 (C-O-C), no spiroketal band; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ1.76 (br s, Me-rha), δ3.24 (3H, s, C-22 OMe), 4.25 (d, 1H, J = 7.1 Hz), 4.57 (d, 1H, J = 7.0 Hz), 5.20 (1H, br s); [α]<sub>D</sub> -84.18° (pyridine; c 1); [M]<sub>D</sub> -771.1° (-778°). (Found: C, 59.98 H, 8.39, C<sub>46</sub>H<sub>76</sub>O<sub>18</sub> requires: C, 60.20; H, 8.29%)

Trigofoenoside A. Amorphous powder,  $R_f$  0.77 (system a), mp 219–221° (decomp.),  $[\alpha]_D$  – 90.1° (pyridine, c 1). FAB-MS m/z: 1036  $[M+H+CS]^+$ , 925  $[M+Na]^+$ , 941  $[M+K]^+$ ; M, 902 for  $C_{45}H_{74}O_{18}$ .

Trigofoenoside D-1. An amorphous solid from MeOH-Me<sub>2</sub>CO,  $R_f$  0.67 (solvent system a), mp 250-253°; IR ν KBr cm<sup>-1</sup>: 3600-3250 (OH), no spiroketal; <sup>1</sup>H NMR (CD<sub>3</sub>OD-CDCl<sub>3</sub>): δ 1.67 (br s, Me-rha), 3.12 (s, C-22 OMe), 4.21 (d, 1H, J = 7.3 Hz), 4.56 (d, 1H, J = 7.5 Hz), 4.92 (d, 1H, J = 7.8 Hz), 5.10 (1H, br s);  $[\alpha]_D - 77.6^\circ$  (pyridine; c 1);  $[M]_D - 836.52^\circ$  (-844°). (Found: C, 58.0, H; 7.82, C<sub>52</sub>H<sub>86</sub>O<sub>23</sub>, requires C: 57.88, H: 7.97.)

Trigofoenoside D. Amorphous powder,  $R_f$  0.55 (system a), mp 246–248° (decomp);  $[\alpha]_D$  – 73.2° (pyridine; c 1).

Enzymatic hydrolysis. Compounds A-1 and D-1 (100 mg) were dissolved in  $H_2O$  (5 ml) and emulsin (almond, 10 ml) soln and one drop of toluene were added to each soln. Mixtures were incubated at 37° for 96 hr before being extracted with n-BuOH and checked by TLC. Water layers were coned and subjected to PC (system c) and TLC (system b). The n-BuOH concentrate from A-1 gave a single Liebermann-Burchard reagent positive spot and recrystallized from MeOH- $H_2O$  as an amorphous powder (PA) mp 198-201° (decomp.);  $[\alpha]_D - 94.7$ ° (pyridine; c 1);  $R_f$  0.87 (system a); IR  $v_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3600-3200 (OH), 985, 920, 900, 850 [920 > 900 (255), spiroketal]. PA on silica gel TLC (system a) gave a superimposable spot with Pro A of dioscin and on complete hydrolysis furnished D-glucose, L-rhamnose (1:1) and yamogenin.

The n-BuOH concentrate of D-1 was crystallized from MeOH. It gave a single Liebermann-Burchard reagent positive spot,  $R_f$  0.70 (system a), mp 301-304° (decomp.);  $[\alpha]_D = 86.2^\circ$  (pyridine; c 1); IR  $v_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3600-3200 (OH), 985, 920, 900, 850 [920 > 900 (25S), spiroketal]; PD on TLC exhibited a superimposable spot with gracillin and on complete hydrolysis furnished D-glucose, L-rhamnose (2:1) and yamogenin.

Identification of aglycone and sugars. Compounds A-1 (80 mg) and D-1 (100 mg) in 2 N HCl ( $\rm H_2O$ -dioxan, 1:1) were refluxed separately for 4 hr. The ppts were collected and purified by crystallization from Me<sub>2</sub>CO-MeOH to afford colourless needles  $R_f$  0.31 (system h), mp 201°; MS m/z: 414 [M]<sup>+</sup>, 139 (base peak) for  $\rm C_{27}H_{42}O_3$ ; [ $\alpha$ ]<sub>D</sub> -129° (CHCl<sub>3</sub>; c 1); IR  $v_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 3600-3200 (OH), 1460, 1380, 980, 921, 898, 861 [921 > 898 (25S), spiroketal]. Co-TLC (system g) on AgNO<sub>3</sub>-impregnated plate (2%), three-fold development, gave superimposable spot with yamogenin.

Each filtrate was neutralized with resin (Dowex-3, OH<sup>-</sup> form) and evaporated to dryness in vacuo. Each residue was examined by PC (system c), TLC (system b) and GLC (condition a). Sugars were identified as glucose ( $R_G$  1.00,  $R_f$  0.08) and rhamnose ( $R_G$  2.00,  $R_f$  0.30) in both. GLC of sugar samples as trimethyl silyl derivatives  $R_t$  (min): glucose (26.9, 28.9), rhamnose (18.4, 19.7).

The molar ratio of sugars was determined by the help of GLC and calorimetry [7] (phenol-H<sub>2</sub>SO<sub>4</sub>), which revealed the proportions of glucose and rhamnose to be 2:1 and 3:1 for A-1 and D-1, respectively.

Methylation of A-1 and D-1. A-1 (75 mg) and D-1 (100 mg) were methylated by Hakomori's method [5] and worked up as usual. The permethylates were obtained as brown residues. Methanolysis with 3% methanolic HCl gave methyl 2,3,4,6-tetra-O-methyl-p-glucopyranoside ( $R_{\rm TMG}$  1.00,  $R_f$  0.61) methyl 2,3,4-tri-O-methyl-1-rhamnopyranoside ( $R_{\rm TMG}$  1.01,  $R_f$  0.67) and methyl 3,4,6-tri-O-methyl-p-glucopyranoside ( $R_{\rm TMG}$  0.78,  $R_f$  0.55) for A-1 and methyl 2,3,4,6-tetra-O-methyl-p-glucopyranoside, methyl 2,3,4-tri-O-methyl-1-rhamnopyranoside and methyl 4,6-di-O-methyl-p-glucopyranoside ( $R_{\rm TMG}$  0.46,  $R_f$  0.38) for D-1, identified by PC after hydrolysis (system f), TLC (system d) and GLC (condition b) with the help of authentic samples.

Periodate treatment. Compounds A-1 and D-1 (30 mg each) were taken in H<sub>2</sub>O (5 ml each) and treated with 0.05 M sodium-m-periodate soln (3 ml) in aq. MeOH. The reaction mixtures were kept in the dark for 48 hr at room temp before being extracted with n-BuOH. The extracts were coned and completely hydrolysed. On usual work up both saponins furnished yamogenin (co-TLC and mmp) and sugar residues which were subjected to PC (system c). No sugar was detected in compound A-1, but D-1 showed the presence of p-glucose.

Partial hydrolytic studies of PA. PA (30 mg) was subjected to partial hydrolysis with 0.1 N HCl in dioxan– $H_2O$  (1:1) for 40 min. It gave a prosapogenin PA<sub>1</sub> (12 mg) and an aglycone after purification by PLC (system b). PA<sub>1</sub> (7 mg),  $R_f$  0.89 (system a), mp 271–273° (decomp.); IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 3600–3250 (OH), 981, 918, 900 and 861 (918 > 900);  $[\alpha]_D$  – 101.6° (pyridine; c 0.5); co-TLC (system b) with trillin exhibited a superimposable spot with PA<sub>1</sub> and on complete hydrolysis PA<sub>1</sub> gave D-glucose and yamogenin as an aglycone, identified with the help of authentic samples.

Partial hydrolytic studies of PD. PD (32 mg) on partial hydrolysis with 0.1 N HCl in dioxan-H<sub>2</sub>O (1:1) for 40 min yielded three prosapogenins namely PD<sub>1</sub>, PD<sub>2</sub> and PD<sub>3</sub> in the order of increasing polarity. PD<sub>1</sub> and PD<sub>2</sub> were found to be

identical to PA<sub>1</sub> and PA as on TLC (system b) both produced similar spots. These were confirmed by mmp, co-IR and complete hydrolysis of each. PD<sub>3</sub>,  $R_f$  0.72 (system b), mp 269–272° (decomp.);  $[\alpha]_D$  –88.3°. On complete hydrolysis D-glucose was the only sugar detected on PC (system c) along with yamogenin in the molar ratio of 2:1.

C-22 hydroxy and C-22 methoxy derivatives. Compounds A-1 and D-1 (30 mg each) were boiled with Me<sub>2</sub>CO-H<sub>2</sub>O (7:3) for 21 hr and kept overnight. After evaporation of the solvent (in vacuo) an amorphous powder was obtained from each compound. The <sup>1</sup>H NMR spectra of both saponins exhibited no methoxy signal. When these products (A and D) were refluxed with dry MeOH for 11 hr, A-1 and D-1 regenerated.

Acknowledgements—The authors thank Dr. Akhtar Husain, Director, CIMAP, for his keen interest and providing necessary facilities for this work, and to Dr. Y. Itagaki, Jeol Ltd., (Japan) for FAB-MS. One of the authors (RKG) thanks CSIR for the award of a research fellowship.

### REFERENCES

- Gupta, R. K., Jain, D. C. and Thakur, R. S. (1984) Phytochemistry 23, 2605.
- Kiyosawa, S. and Hutoh, M. (1968) Chem. Pharm. Bull. 16, 1162.
- Rothman, E. S., Wall, M. E. and Eddy, C. R. (1952) J. Am. Chem. Soc. 74, 4013.
- Schulten, H.-R., Singh, S. B. and Thakur, R. S. (1984) Z. Naturforsch 39C, 201.
- 5. Hakomori, S. (1964) J. Biochem. 55, 205.
- 6. Klyne, W. (1950) Biochem, J. 47, 12.
- Dubois, M., Gillis, K., Hamilton, J. K., Rabbers, P. A. and Smith, F. (1956) Analyt. Chem. 28, 350.